

Arsenic RBA Study



DEPARTMENT OF TOXIC
SUBSTANCES CONTROL

Objective of the research

Provide better tools to assess health risks at MSL that allow use of bioavailability in risk assessment and risk management decisions by:

- Developing cost effective methods to determine arsenic bioavailability:
 - Improve correlation between in vitro and in vivo methods through improvement of the in vitro simulated gastro-intestinal assay
 - Identify geochemical and mineralogical parameters which control RBA of soil-bound As, and inexpensive bench procedures for estimation of RBA acceptable in a regulatory setting.

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Objective cont.

- Identifying wet chemical, spectroscopic, and physical measurements to use in characterizing MSL
- Developing a database of mine wastes and corresponding in vitro and mineralogy data
- Establishing a methodology for implementation at sites other than EMSHP
- Developing a guidance document for use of bioavailability at MSL

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First Sampling Event

- Sampling done 9/21-9/23
- Total of 25 individual samples in 46 five gallon containers collected
- As concentrations ranged from 9-9,700 ppm
- Sample processing at OSU prior to distribution to investigators

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Dr. Valerie Mitchell with HERO presented a poster entitled “Identifying Predictors for Bioavailability of Arsenic in Soil at Mining Sites” at the 49th Annual Meeting of the Society of Toxicology in March 2010.

Has submitted an abstract for a poster presentation at SOT in March 2011.

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Soil Processing at OSU

- Homogenization by tumbling for an extended period in a mixer.
- Sieved on a shaker table and the 250 μm fraction is split and tested for homogeneity.
- Homogeneity evaluation includes taking a total of 24 subsamples from each homogenized sample and testing each subsample for total arsenic using USEPA Method 3051a.
- Need ~ 3.5 kgs of 250 μm fraction for each sample. Most samples yielding 15-20 % but a few were <5%.

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Synchrotron and QEMSCAN

- Thin sections prepared from rock samples collected by T. Burlak during Sampling Event 1 were analyzed at SSRL and University of Utah.
- Purpose is to understand the progression of weathering, particularly the role of Fe-Mg carbonate minerals in controlling the type of Fe (hydr)oxide that forms when arsenic-bearing sulfides weather.

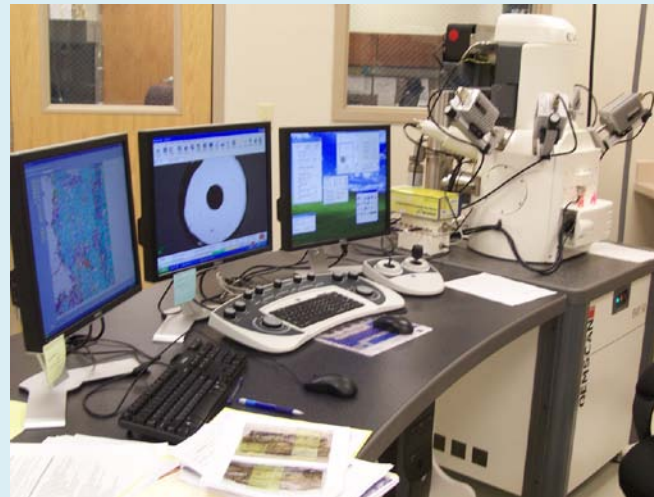
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“An SEM on Steroids”

- Four high-speed energy dispersive detectors collect and combine element x-ray data to identify minerals
- Thousands of high-res images can be collected very rapidly
- Images created by collecting 1000-count EDAX (Energy Dispersive X-Ray Analysis) spectra on each analysis spot
- Scanning a typical thin section can take a couple hours depending on the resolution
- Can detect fine grained materials down to 1 micron
- Statistical evaluation of data used to quantitatively characterize rock mineralogy (95% confidence interval)



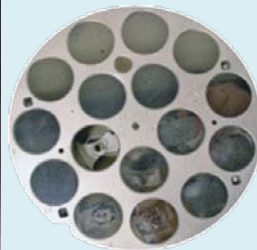
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Carbon Coated Thin Sections Loaded into Stage Holder

- Carbon coat to help prevent stray X-rays from sampling nearby areas and interfering with analysis of desired point
- Stages available for round epoxy plugs as well as thin sections
- Thin section stage includes gold and quartz samples so beam can recalibrate every 30-40 mins or so



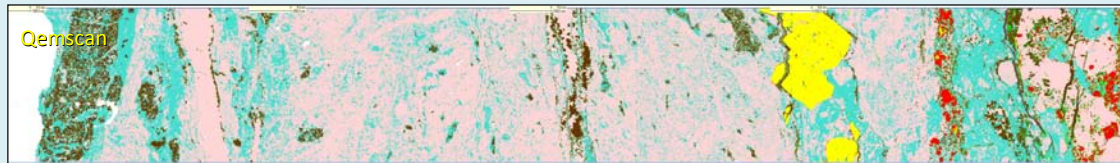
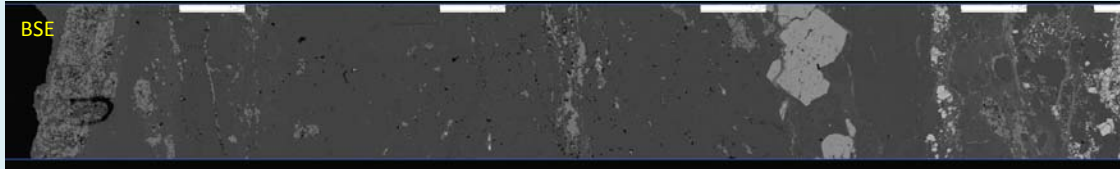
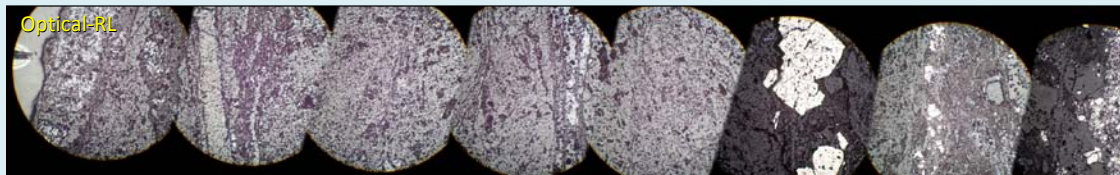
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A6-B-B

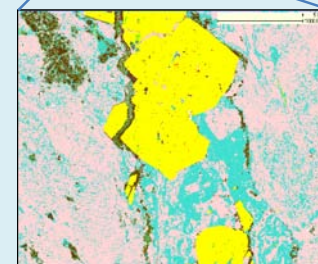
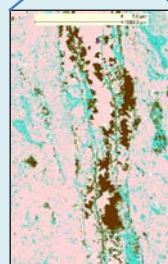
Erich Petersen, University of Utah



5 μm - 22x3 mm



Quartz	56.23
Feldspar	24.20
Fe Oxides(G07)	4.68
Other_oxides	3.96
Chlorite	3.53
Pyrite	3.37
Arsenopyrite	1.77
Element_Map-As	1.21
Background	0.90
Other_Silicates	0.85
Ca/Sr/SO4/Gypsum	0.30
Others	0.12



V2.0

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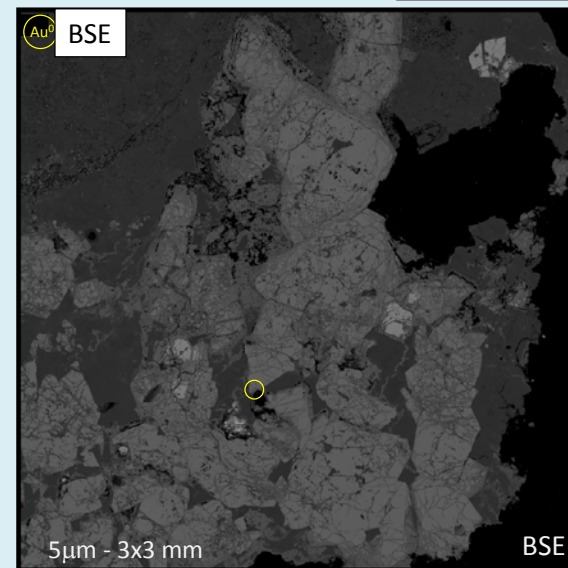
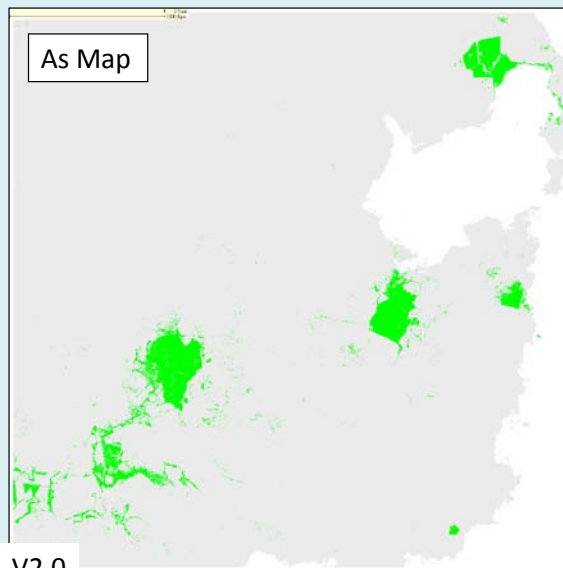


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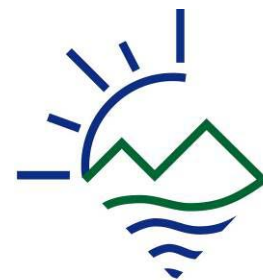
A10-G-C

Erich Petersen, University of Utah

The As Map shows in green all areas that contained measurable As. The As-rich areas consist of arsenopyrite (see next slides) and As-rich iron oxides that are forming due to the weathering of arsenopyrite.



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Questions